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         FEB 23
                 TOXCENTER updates mirror those of MEDLINE - more
                 precise author group fields and 2009 MeSH terms
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         FEB 23
                 Three million new patent records blast AEROSPACE into
                 STN patent clusters
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         FEB 25
                 USGENE enhanced with patent family and legal status
                 display data from INPADOCDB
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         MAR 06
                 INPADOCDB and INPAFAMDB enhanced with new display
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                 applications and grants
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                 ESBIOBASE reloaded and enhanced
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                 for nanomaterial substances
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                 CAS coverage of exemplified prophetic substances
                  enhanced
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                  information
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         APR 26
                 USPATFULL and USPAT2 enhanced with patent
                 assignment/reassignment information
NEWS 25
         APR 28
                 CAS patent authority coverage expanded
NEWS 26
         APR 28
                 ENCOMPLIT/ENCOMPLIT2 search fields enhanced
NEWS 27
         APR 28
                 Limits doubled for structure searching in CAS
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(HOPS OR HOPSES)
=> s 12 and polyphenols
         20961 POLYPHENOLS
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           134 L2 AND POLYPHENOLS
=> s 13 and extract
         54659 EXTRACT
         55277 EXTRACTS
        105133 EXTRACT
                 (EXTRACT OR EXTRACTS)
        372091 EXT
        251184 EXTS
        554848 EXT
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        591656 EXTRACT
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            29 L4 AND PY<=2003
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L6 ANSWER 1 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2006:53217 CAPLUS

DOCUMENT NUMBER: 144:135253

TITLE: Pharmaceutical compositions of hops resins

INVENTOR(S): Kuhrts, Eric H.

PATENT ASSIGNEE(S): USA

SOURCE: U.S. Pat. Appl. Publ., 13 pp., Cont.-in-part of U.S.

Ser. No. 140,495. CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT	PATENT NO.				KIND DATE			APPLICATION NO.						DATE			
US 2006	0013	 870		A1	A1 20060119		1	US 2005-152023					20050613				
US 20030228369				A1 20031211			1	US 2002-140495						20020506 <			
AU 2006259561				A1	A1 20061228				AU 2006-259561						20060612		
CA 2611898				A1 20061228				(	CA 2006-2611898						20060612		
WO 2006138253				A1 20061228			1	WO 2	006-1	JS22	886		20060612				
W:	ΑE,	AG,	AL,	AM,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,	
	CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FΙ,	GB,	GD,	
	GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	KM,	KN,	KP,	KR,	
	KΖ,	LC,	LK,	LR,	LS,	LT,	LU,	LV,	LY,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	
	ΜZ,	NA,	NG,	ΝΙ,	NO,	NΖ,	OM,	PG,	PH,	PL,	PT,	RO,	RS,	RU,	SC,	SD,	
	SE,	SG,	SK,	SL,	SM,	SY,	ΤJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	
	VC,	VN,	ZA,	ZM,	ZW												
RW:	ΑT,	BE,	ВG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	
	IS,	ΙΤ,	LT,	LU,	LV,	MC,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,	BF,	ВJ,	
	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,	SN,	TD,	TG,	BW,	GH,	

GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,

KG, KZ, MD, RU, TJ, TM

PRIORITY APPLN. INFO.: US 2002-140495 A2 20020506

> US 2005-152023 A 20050613

> WO 2006-US22886 W 20060612

The present invention concerns a pharmaceutical composition comprising a dry AB free flowing powder. The powder can include various combinations of alpha acid, iso-alpha acids, and beta acids. The composition can further include a silica salt absorbent and/or an antioxidant. These compns. are preferably prepared by mixing hops ext. with an absorbent in a high intensity mixer without added solvent. Thus, a hops resin containing 60 wt % alpha acids was converted into a powder by using hops resin 68, calcium silicate 25, maltodextrin 5.5, and ascorbic acid 1.5%.

ANSWER 2 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2005:687088 CAPLUS

DOCUMENT NUMBER: 143:438974

TITLE: Replacement of kettle hops with (reduced)

> isomerised hop extracts: implications for beer bitterness and flavour stability

AUTHOR(S):

De Cooman, Luc; Aerts, Guido; De Rouck, Gert; Syryn, Evelien; Van Opstaele, Filip; Goiris, Koen; De Ridder,

Marjan; Joos, Pieter; De Keukeleire, Denis

CORPORATE SOURCE: Laboratory of Enzyme and Brewing Technology, KaHo

St.-Lieven, Ghent, B-9000, Belg.

Proceedings of the Congress - European Brewery SOURCE:

Convention (2003), 29th, 24/1-24/13

CODEN: EBCPA6; ISSN: 0367-018X

PUBLISHER: Fachverlag Hans Carl GmbH

DOCUMENT TYPE: Journal; (computer optical disk)

LANGUAGE: English

Pre-isomerized hop exts. do not contain polyphenols.

Thus, their application could adversely affect flavor stability. This study on similar pilot beers demonstrates that exclusive bittering with  $iso-\alpha$ -acids during wort boiling is not to the detriment of flavor stability. In comparison with conventional pellet hopping, advanced bittering is at least 'neutral' to flavor stability and markedly pos. in respect of bitterness quality. Flavor deterioration is further delayed when tetrahydroiso- $\alpha$ -acids are included post-fermentation. These observations raise the question on the relevance of the reducing power of kettle hops. Advanced hopping presents an alternative in view

of refined bitterness and enhanced flavor stability.

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 3 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

2002:941421 CAPLUS ACCESSION NUMBER:

138:72284 DOCUMENT NUMBER:

TITLE: Xanthohumol in beer - possibilities and limitations of

enrichment

Forster, A.; Gahr, A.; Ketterer, M.; Beck, B.; AUTHOR(S):

Massinger, S.

CORPORATE SOURCE: Nateco GmbH & Co. KG, St. Johann/Hallertau, Germany

SOURCE: Monatsschrift fuer Brauwissenschaft (2002),

55(9/10), 184-186, 188-194

CODEN: MOBRDJ; ISSN: 0723-1520

PUBLISHER: Fachverlag Hans Carl

DOCUMENT TYPE: Journal LANGUAGE: German

Xanthohumol, a polyphenol of hops, is said to have pos. physiol.

properties. Adequate research particularly on living cells has not been completed for a long time yet, but it still might be of interest, if and how xanthohumol may be enriched in beer. Besides conventional pellets or spent hops after CO2-extraction there are already especially developed xanthohumol exts. available, which may be specifically used. However, there are limits to elevation of xanthohumol levels and its isomer isoxanthohumol in com. filtered beers. Thus, xanthohumol and resp. isoxanthohumol levels in, for example bottom fermented com. beers, differ on a low scale from less than 0.1 ppm or rather less than 1 ppm. At the end of the production process of beer there can only 10-20% rel. be found of the amount of xanthohumol that has been spiked to wort. If xanthohumol is supposed to be enriched effectively a dosage with special xanthohumol exts. after fermentation or yeast filtration is recommended, whereas a constant turbidity has to be anticipated. Thus, beers rich in xanthohumol/isoxanthohumol are currently only imaginable as specialties. Beyond that, xanthohumol exts. can be added to any kind of turbid and lightly bitter beverages, which on their part can be the basis of beer mix beverages.

REFERENCE COUNT: THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS 16 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 4 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2002:326574 CAPLUS

DOCUMENT NUMBER: 136:306887

Process for producing tannins from plants TITLE: INVENTOR(S): Kim, Yong Wook; Lim, Se Jin; Kim, Myoung Ae
PATENT ASSIGNEE(S): S. Korea
SOURCE: Repub. Korean Kongkae Taeho Kongbo, No pp. given

CODEN: KRXXA7

DOCUMENT TYPE: Patent Korean LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE A 20000915 KR 1999-4505 19990209 <--KR 1999-4505 19990209 KR 2000055727 PRIORITY APPLN. INFO.:

A process of extracting tannins and polyphenol components from plants is provided. This process comprises: (1) extracting the skin of chestnut, persimmon, pine needle and hops with acetone, filtering and removing acetone to give crude exts., (2) removing water from the crude exts. and separating tannin components in a low polar organic solvent (petroleum ether, ether, Et acetate and butanol); said tannin components are gallic acid derivs., ellagic acid and catechin.

ANSWER 5 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:780207 CAPLUS

DOCUMENT NUMBER: 137:19527

TITLE: Accelerated solvent extraction in the investigation of

polyphenols in the brewing process Papagiannopoulos, M.; Mellenthin, A.

AUTHOR(S): CORPORATE SOURCE: Institut fur Lebensmittelwissenschaft und

Lebensmittelchemie, Rheinische

Friedrich-Wilhelms-Universitat Bonn, Bonn, 53115,

Germany

SOURCE: Special Publication - Royal Society of Chemistry (

2001), 269 (Biologically-Active Phytochemicals

in Food), 199-201

CODEN: SROCDO; ISSN: 0260-6291

PUBLISHER: Royal Society of Chemistry DOCUMENT TYPE: Journal LANGUAGE: English

AB The use of accelerated solvent extraction for the anal. of polyphenols in hops, malt and samples of the brewing process was evaluated.

An optimized sample cleanup for subsequent HPLC-MS/MS anal. was also investigated. Extns. were carried out on an Automated Sample Extractor equipped with a Solvent Delivery Module for the use of solvent mixts. ASE offers a high potential for the anal. of phenolic compds. from solid sample materials. Extraction efficiency is higher compared to manual extraction

with a decreased amount of matrix interferences. ASE delivers exts . with a higher concentration of desired analytes in a reduced volume of solvent

used. Moreover, extraction is much faster and allows for a higher number of analyses to be carried out in a given time. There is a reduced need for subsequent time consuming steps, like concentration by solvent evaporation, thus

minimizing the possibility of alteration and degradation of sample compds.

L6 ANSWER 6 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:369105 CAPLUS

DOCUMENT NUMBER: 135:210293

TITLE: Reducing power of various hop varieties AUTHOR(S): Lermusieau, G.; Liegeois, C.; Collin, S.

CORPORATE SOURCE: Unite de Brasserie et des Industries Alimentaires,

Universite Catholique de Louvain, Louvain-la-Neuve,

B-1348, Belg.

SOURCE: Cerevisia (2001), 26(1), 33-41 CODEN: CEREFI; ISSN: 1373-7163

PUBLISHER: Cerevisia
DOCUMENT TYPE: Journal
LANGUAGE: English

Since lipid autoxidn. during wort boiling is determinant for the appearance of staling flavor in aged beers, the reducing power of hops added in the boiling kettle was investigated. An assay based on the inhibition of linoleic acid oxidation in the presence of an initiator (2,2'-azobis(2-amidino-propane) dihydrochloride; AAPH) enabled us to distinguish hop varieties and conditionings. Large differences in hop flavanoid contents explained the higher antioxidant activity of  $low-\alpha$ -acid samples vs. bitter varieties and CO2 hop exts. When the  $\alpha$ -acids reducing power was subtracted, very good correlation was observed between the resulting inhibition time and the amount of total flavanoids in pellets. An anal. of the hop polyphenols content explains this result since flavanoids represent more than 80% of phenolic compds. As expected, adding hop pellets to the kettle effectively increased the overall reducing activity of wort. Supercrit. CO2 hop exts. had no significant effect due to their extremely low level of polyphenols. The concentration of the very well-known marker of beer ageing, trans-2-nonenal, was lower in wort boiled with hop exhibiting a better reducing power.

REFERENCE COUNT: 32 THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 7 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:109514 CAPLUS

DOCUMENT NUMBER: 134:352554

TITLE: Reducing power of hop cultivars and beer ageing

AUTHOR(S): Lermusieau, G.; Liegeois, C.; Collin, S.

CORPORATE SOURCE: Unite de Brasserie et des Industries Alimentaires, Universite Catholique de Louvain, Louvain-la-Neuve,

B-1348, Belg.

SOURCE: Food Chemistry (2001), 72(4), 413-418

CODEN: FOCHDJ; ISSN: 0308-8146

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

Since lipid autoxidn. during wort boiling is a factor in the development AB of staling in aged beers, the authors investigated the reducing power of hops added in the boiling kettle. An assay based on the inhibition of linoleic acid oxidation in the presence of an initiator [2,2'-azobis(2-amidino-propane) dihydrochloride = AAPH] enabled the authors to distinguish hop varieties and conditionings. Large differences in hop flavanoid contents explained the higher antioxidant activity of  $low-\alpha$ -acid samples vs. bitter varieties and CO2 hop exts. As expected, adding hop pellets to the kettle effectively increased the overall reducing activity of wort. Supercrit. CO2 hop exts. had no significant effect due to their extremely low level of polyphenols. The concentration of the very well-known marker of beer ageing, trans-2-nonenal, was lower in boiled wort exhibiting a better reducing power. The AAPH reducing power test applied to hops or worts was thus efficient in predicting nonenal synthesis during boiling. Hop varieties and conditionings emerged from this work as key parameters

final product.

REFERENCE COUNT: 22 THERE ARE 22 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

for improving the reducing power of wort and the flavor stability of the

L6 ANSWER 8 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:575706 CAPLUS

DOCUMENT NUMBER: 133:192167

TITLE: Hops

AUTHOR(S): Yoshida, Manabu

CORPORATE SOURCE: Res. Inst. New Prod. Dev., Suntory Ltd., Japan SOURCE: Nippon Jozo Kyokaishi (2000), 95(8), 550-559

CODEN: NJKYES; ISSN: 0914-7314

PUBLISHER: Nippon Jozo Kyokai DOCUMENT TYPE: Journal; General Review

LANGUAGE: Japanese

AB A review with 18 refs., on the major cultivation area of Humulus lupulus, biosynthesis of  $\alpha$ - and  $\beta$ -acids and their properties, essential oils and phenolic components of hops, pharmacol. effects of hop polyphenols, roles of hops in beer brewing, application of hop products (hop pellet, hop exts., iso- $\alpha$ -acid, modified iso $\alpha$ -acid, etc.), and anal. of bitter components of hop products.

L6 ANSWER 9 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:492946 CAPLUS

DOCUMENT NUMBER: 133:149724

TITLE: Possibilities of using glucose-maltose syrups as

substitutes for wort extract

AUTHOR(S): Smogrovicova, Daniela; Domeny, Zoltan; Patkova,

Jaroslava; Bafrncova, Petra

CORPORATE SOURCE: Katedra Biochem. Technologie, Chemickotechnol.

Fakulta, Slovenska Tech. Univ., Bratislava, Slovakia

SOURCE: Kvasny Prumysl (2000), 46(5), 133-136

CODEN: KVPRAB; ISSN: 0023-5830

PUBLISHER: Vyzkumny Ustav Pivovarsky a Sladarsky

DOCUMENT TYPE: Journal LANGUAGE: Slovak

AB The use of Glucoplus 830 (containing mainly glucose and maltose) and

Fermentose 352 (containing mainly maltose) made by the Amylum Co. (Slovakia)

as 10-50% substitutes for wort ext. in 12° beer brewing was studied under laboratory conditions in 2-L batches. Sucrose substitution was used for comparison. The sugar substitutes were added before boiling with hops and fermentation with Saccharomyces cerevisiae yeasts. The one-phase fermentation lasted 12 days at 10°C. The produced beers were evaluated for bitterness and color and chemical analyzed for pH, total polyphenols, total and protein N, apparent and true ext ., ethanol, volatile compds., and diacetyl. The sugar substitutes affected the beer taste, flavor and foam stability especially at higher doses used. The practical impact of these changes would require a testing on a full production scale.

L6 ANSWER 10 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1997:184419 CAPLUS

DOCUMENT NUMBER: 126:176658

ORIGINAL REFERENCE NO.: 126:34009a,34012a

TITLE: polyphenols from hops for

manufacturing cosmetics or other products

INVENTOR(S): Tagashira, Motoyuki; Uemitsu, Nobuo

PATENT ASSIGNEE(S): Asahi Breweries Ltd, Japan SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 09002917	A	19970107	JP 1995-173931	19950619 <
JP 3477628	В2	20031210		
PRIORITY APPLN. INFO.:			JP 1995-173931	19950619

AB Polyphenols from exts. of hops for manufacturing

cosmetics or other products are claimed. Thus, skin-care hand lotions were prepared containing carbowax 1500 15, ethanol 8, propylene glycol 90, water

52.5, the polyphenols 0.2 weight parts, perfumes, and preservatives.

L6 ANSWER 11 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:405857 CAPLUS

DOCUMENT NUMBER: 111:5857

ORIGINAL REFERENCE NO.: 111:1151a,1154a

TITLE: Protein precipitation during wort boiling: quality

aspects of diminished wort boiling times of brews prepared from proanthocyanidin-free or regular raw

materials

AUTHOR(S): Delcour, Jan A.; Vanhamel, Sonja; Moerman, Etienne;

Vancraenenbroeck, Roger

CORPORATE SOURCE: Lab. Toegespaste Org. Scheikd., Kathol. Univ. Leuven,

Heverlee, B-3030, Belg.

SOURCE: Journal of the Institute of Brewing (1988),

94(6), 371-4

CODEN: JINBAL; ISSN: 0368-2587

DOCUMENT TYPE: Journal LANGUAGE: English

AB When using proanthocyanidin-free materials for the production of beer, a red. of the wort boiling time can be considered. In worst prepared with regular malt and tannin-free hop ext. there is a continuous precipitation of the malt flavanoids while in brews prepared from a proanthocyanidin-free malt and regular hops there is a simultaneous extraction and removal of

the hop flavanoids leading to constant levels of these hop flavanoids. The results also show that the level of Kjeldahl nitrogen in wort boiled with hops will be the same as that in worts boiled with n-hexane tannin-free hop exts. These results and the fact that more protein ppts. in brews containing no malt or hop proanthocyanidins suggest that, unlike what is the case during the development of beer haze, polyphenols are not necessary for an effective protein precipitation during wort boiling.

L6 ANSWER 12 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1988:185189 CAPLUS

DOCUMENT NUMBER: 108:185189

ORIGINAL REFERENCE NO.: 108:30405a,30408a

TITLE: Interactions of polyphenols with proteins

during hop boiling

AUTHOR(S): Skach, Josef; Mikyska, Alexandr

CORPORATE SOURCE: Vyzk. Ustav Pivovar. Sladarsky, Prague, Czech.

SOURCE: Kvasny Prumysl (1987), 33(8-9), 251-4

CODEN: KVPRAB; ISSN: 0023-5830

DOCUMENT TYPE: Journal LANGUAGE: Czech

AB The effect of polyphenol compds. of malt and hops on protein

precipitation during hop boiling was tested on a laboratory scale. Polyphenol

exts. of hops consist of complex polyphenols

and proteins that are heat-resistant. Wort consists of a large quantity

of single polyphenols which are resistant to polymerization and

reactions with proteins during boiling. Due to the simultaneous effects

of malt and hop polyphenols that complex with proteins, a

significant quantity of proteins is precipitated in the form of tannin-protein

complexes during boiling.

L6 ANSWER 13 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1988:73688 CAPLUS

DOCUMENT NUMBER: 108:73688

ORIGINAL REFERENCE NO.: 108:12191a,12194a

TITLE: Effect of hop polyphenols on the

physicochemical characteristics and colloidal

stability of beer

AUTHOR(S): Aksenova, Z. N.; Linetskaya, G. N.; Budko, L. V.;

Fedorova, S. S.; Kozhukhar, M. M.; Shmuilovich, D. S.

CORPORATE SOURCE: NPO Napitkov Miner. Vod, Kharkov, USSR

SOURCE: Fermentnaya i Spirtovaya Promyshlennost (1987

), (6),  $3\overline{5}$ -7

CODEN: FSPMAM; ISSN: 0367-3197

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB A correlation was found between the concns. of polyphenols and

anthocyanogens and the quality of beer and worts. Chemical anal. showed that

the major source of polyphenols in beer and worts is barley and

malt rather than hops. Addition of hop ext. to worts had no effect on the levels of polyphenols, the physicochem.

properties, or the colloidal stability of beer.

L6 ANSWER 14 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1986:87057 CAPLUS

DOCUMENT NUMBER: 104:87057

ORIGINAL REFERENCE NO.: 104:13813a,13816a

TITLE: Semitechnical brewing assay for estimation of tannin

constituents of hops and hop products

AUTHOR(S): Puespoek, J.

CORPORATE SOURCE: Austria

SOURCE: Mitteilungen der Versuchsstation fuer das

Gaerungsgewerbe in Wien (1985), 39(9-10),

120 - 5

CODEN: MVGGAN; ISSN: 0369-271X

DOCUMENT TYPE: Journal LANGUAGE: German

AB The properties of beers that had been made with various hop products are

described. The original ext. had the highest tannin content and

precipitated proteins the most, resulting in the poorest foam. The color was

only slightly affected by the original ext. Colloidal stability

was improved when an ext. lacking polyphenols was

used. Beer flavor was not much affected, but flavor stability was bad

when a polyphenol-poor hop ext. was used.

L6 ANSWER 15 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1985:594901 CAPLUS

DOCUMENT NUMBER: 103:194901 ORIGINAL REFERENCE NO.: 103:31393a

TITLE: Variations of alpha and iso-alpha-compounds of

hops in packaged beer during various storage

conditions

AUTHOR(S): Jaeger, P. CORPORATE SOURCE: Austria

SOURCE: Mitteilungen der Versuchsstation fuer das

Gaerungsgewerbe in Wien (1985), 39(7-8),

88-101

CODEN: MVGGAN; ISSN: 0369-271X

DOCUMENT TYPE: Journal LANGUAGE: German

AB Bottled beers that had been made with hop pellets or various hop exts. were stored in the dark or under lights for ≤9 wk at

temps. between 10° and 40°. Diverse hop products all showed

reproducible changes in their iso- $\alpha$ -acid composition. There is a close

correlation between the effects of light and temperature on the

iso-lpha-acids and on the polyphenols or anthocyanogens of

beer. This explains the effects of storage on both the colloidal

stability and flavor stability of beer.

L6 ANSWER 16 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1984:137293 CAPLUS

DOCUMENT NUMBER: 100:137293

ORIGINAL REFERENCE NO.: 100:20937a,20940a

TITLE: Tanning properties of flavanols in barley and

hops measured by reaction with cinchonine

sulfate in relation to haze formation in beer

AUTHOR(S): McMurrough, I.; Hennigan, G. P.

CORPORATE SOURCE: Res. Lab., Arthur Guinness Son and Co. (Dublin) Ltd.,

Dublin, 8, Ire.

SOURCE: Journal of the Institute of Brewing (1984),

90(1), 24-32

CODEN: JINBAL; ISSN: 0368-2587

DOCUMENT TYPE: Journal LANGUAGE: English

AB The flavanoid polyphenol exts. from barley and hops

were each separated into 6 fractions by adsorption chromatog. on Sephadex

LH20. These fractions were further characterized by several anal.

methods, including high-performance liquid chromatog. and a colorimetric measurement of polymerization index. The tanning powers of the fractions were graded according to their reactivities with cinchonine sulfate solution in a standardized turbidometric test. Whereas almost 75% of the flavanols from Ark Royal barley were non-tanning oligomers, almost 96% of the flavanols

from Bullion hops were polymeric tannins. Reactivity of most of the barley flavanols with cinchonine sulfate was increased greatly by oxidation with peroxidase and H2O2. Some effects of polymerization, caused by enzyme action or by exposure to air, on oxidizable polyphenols (nontannins) were measured using (+)-catechin [154-23-4], procyanidin B3 [23567-23-9], and prodelphinidin B3 [78362-05-7] in model systems. These and other measurements on exptl. and com. beers indicated that oxidation of simple flavanols from barley produced polymers with tanning properties. In contrast, the hop flavanols when extracted, apparently in their native forms, were capable of copptg. with polypeptides in beer. Treatment of beers with different stabilizing agents, such as Polyclar AT and silica hydrogel, retarded haze formation by restricting protein-polyphenol interactions.

L6 ANSWER 17 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1978:87596 CAPLUS

DOCUMENT NUMBER: 88:87596

ORIGINAL REFERENCE NO.: 88:13745a,13748a

TITLE: Flavor compounds in barley, malt and wort

AUTHOR(S): Kringstad, Hans K.

CORPORATE SOURCE: Bryggeriind. Forskningslab., Swed. SOURCE: Brygmesteren (1977), 34(10), 215-27

CODEN: BRYGAW; ISSN: 0007-2737

DOCUMENT TYPE: Journal LANGUAGE: Norwegian

When wort without hops is extracted with ether at low pH, substances with a characteristic flavor are extracted Exts. of barley and malt resulting from organic solvents do not give any particular aroma. If these exts. are later treated with H2O aroma occurs. The aroma appears to be formed during mashing. The flavors appear to be hydrolysis products of polyphenols, derivs. of cinnamic acid [621-82-9], particularly ferulic acid [1135-24-6]. Solvents which dissolve polyphenols have high concns. of flavor compds. Part of the flavor compds. remain in the grain and can only be dissolved by a new treatment with H2O. One explanation is that a fraction of the flavor compds. is strongly bound to the protein in the grain. Several extns. were used and the fractions characterized by thin-layer chromatog. and UV spectroscopy. Ferulic acid was predominant in the exts. An acetone extraction of malt upon steam distillation gave an intense characteristic

odor.

L6 ANSWER 18 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1976:41935 CAPLUS

DOCUMENT NUMBER: 84:41935

ORIGINAL REFERENCE NO.: 84:6895a,6898a

TITLE: Variations of polyphenols and their P. I.

[polymerization index] during hop boiling in water and

wort

AUTHOR(S): Narziss, Ludwig; Bellmer, Horst G.

CORPORATE SOURCE: Tech. Univ. Muenchen, Freising-Weihenstephan, Fed.

Rep. Ger.

SOURCE: Brauwissenschaft (1975), 28(11), 332-43

CODEN: BRWSAO; ISSN: 0006-9337

DOCUMENT TYPE: Journal LANGUAGE: German

AB The color, rapid reducing power, and P.I. of the hop polyphenols of the pitching wort increased depending on the polymerization level of the polyphenols in the added hops. A correlation exists between the P.I. of the hop polyphenols and the P.I. of the polyphenols of the pitching wort. The more polymerized the hop

polyphenols are, the lower is their loss during wort boiling and the more highly polymerized polyphenols remain in the pitching wort. When using hops or hop products with known high P.I., an increase in color in the pitching wort results from the increase in color of the polyphenols. When employing tannin-containing hop ext. the color of the reductones also increases. With increasing oxidation of the hop polyphenols, the reducing power of the polyphenols in the pitching wort is decreased. The ratio of polyphenols to color and to reducing power of the pitching wort was determined

L6 ANSWER 19 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1976:41919 CAPLUS

DOCUMENT NUMBER: 84:41919

ORIGINAL REFERENCE NO.: 84:6895a,6898a

TITLE: Polyphenol content and polymerization index of

hops and hop products

AUTHOR(S): Narziss, Ludwig; Bellmer, Horst G.

CORPORATE SOURCE: Tech. Univ. Muenchen, Freising-Weihenstephan, Fed.

Rep. Ger.

SOURCE: Brauwissenschaft (1975), 28(10), 285-92

CODEN: BRWSAO; ISSN: 0006-9337

DOCUMENT TYPE: Journal LANGUAGE: German

AB The polyphenol content and polymerization index (P. I.) was studied during whole

hop processing to a powder or ext. and during storage. Both values in hops and hop powder were subject to seasonal effects. During processing from whole hops to powder, polyphenol content increased and P. I. decreased. Hot water extracted only 70% of the DMF-extractable polyphenols of hops and thus, standard ext. contained only 19-26% of the polyphenols present in whole hops. The P. I. of hop exts. was higher than that of whole hops. Low temperature storage resulted in variable decreases in polyphenol content and produced increases in P. I. in all studied products. High temperature storage resulted in accelerated transformations.

L6 ANSWER 20 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1974:550214 CAPLUS

DOCUMENT NUMBER: 81:150214

ORIGINAL REFERENCE NO.: 81:23443a,23446a

TITLE: Color formation during the brewing process

AUTHOR(S): Narziss, Ludwig

CORPORATE SOURCE: Inst. Technol. Brau. I, Tech. Univ. Muenchen,

Weihenstephan, Fed. Rep. Ger.

SOURCE: Brauwelt (1974), 114(19), 355-64, 366-8

CODEN: BRUWAQ; ISSN: 0724-696X

DOCUMENT TYPE: Journal; General Review

LANGUAGE: German

AB A review with 34 refs. Malt, hop exts., and composition of wort are all responsible for color formation during brewing. The color formation arises through 4 processes: (1) formation ofmelanoidin, (2) enzymic oxidation of malt polyphenols, (3) oxidation ofpolyphenols of hops, and (4) oxidation of reductones.

L6 ANSWER 21 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1974:518491 CAPLUS

DOCUMENT NUMBER: 81:118491

ORIGINAL REFERENCE NO.: 81:18745a,18748a

TITLE: Proportion and composition of nitrogenous substances

in nonbiological beer hazes in relation to the type of

hopping

AUTHOR(S): Mostek, Josef; Cizkova, Hana; Svoboda, Josef CORPORATE SOURCE: Fac. Food Biochem. Technol., Chem.-Technol. Coll.,

Prague, Czech.

SOURCE: Brauwissenschaft (1974), 27(6), 149-55

CODEN: BRWSAO; ISSN: 0006-9337

DOCUMENT TYPE: Journal LANGUAGE: German

AB Beers brewed with CH2Cl2 hop exts. aged more rapidly and produced more haze than beers prepared with natural hops. The haze solids also contained up to 10% more total N and up to 26% more amino

acids; polyphenols and anthocyanogens were decreased.

L6 ANSWER 22 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1973:476909 CAPLUS

DOCUMENT NUMBER: 79:76909

ORIGINAL REFERENCE NO.: 79:12481a,12484a

TITLE: Polyphenols in brewing. II. Fractionation

of polyphenols by ion-exchange

chromatography Sogawa, Hiroshi

AUTHOR(S): Sogawa CORPORATE SOURCE: Japan

SOURCE: Report of the Research Laboratories of Kirin Brewery

Company (1972), No. 15, 17-24 CODEN: RLKBAD; ISSN: 0075-6229

DOCUMENT TYPE: Journal LANGUAGE: English

AB Polyphenols in concentrated exts. of beer were quant. fractionated into several groups by column chromatog. using Dowex 1

+ 4 anion-exchange resin. The column was eluted with 10, 20, 25, and 30% HCO2H in 75% MeOH. The polyphenol fractions were examined by uv absorption, paper chromatog, and differential spectrophotometry; 14 fractions were obtained from one brand of domestic beer. The extent of interaction between protein and polyphenol was strong ( $\Delta$ OD greater than 0.8), medium ( $\Delta$ OD = 0.3-0.8) or slight ( $\Delta$ OD = less than 0.3) and 2 fractions showed strong and 6 showed moderate interaction. Some fractions of polyphenols treated with polyamide (20 g/l. of

beer) were lost and the size of the fractions also decreased. These fractions are probably anthocyanogens. With hops, the major polyphenols were found in some fractions, with lesser amts. in others. Polyphenols in malt differed considerably from those in hops, and the former contributed more to beer polyphenols than did the latter. As with domestic brands, many different chromatog.

patterns were obtained. The contents of some fractions of the polyphenols changed during storage of the beer. Approx. 8% of the polyphenols had a mol. weight of 500, 90% 500-10,000, and only a

trace had a mol. weight >10,000.

L6 ANSWER 23 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1973:430459 CAPLUS

DOCUMENT NUMBER: 79:30459
ORIGINAL REFERENCE NO.: 79:4949a,4952a

TITLE: Methods of determination of polyphenols in

the brewery

AUTHOR(S): Jerumanis, J.

CORPORATE SOURCE: Sect. Brass., Univ. Louvain, Louvain, Belg.

SOURCE: Bulletin de l'Association Royale des Anciens Etudiants

en Brasserie de l'Universite de Louvain (1973

), 69(1), 1-14

CODEN: BAEBA2; ISSN: 0365-8775

DOCUMENT TYPE: Journal LANGUAGE: French

AB After studying several methods of determining polyphenols in brewing products, a method based on use of ammoniacal Fe citrate in alkaline solution

was

considered the most satisfactory. Details of 2 variants of the procedure have been published, and 1 of them was adopted by the European Brewery Convention. However, the other variant may be superior because it is more accurate with darker colored beers. Details of extraction and determination of polyphenols in hops, hop ext., barley, and malt are given. Anthocyanogens represent 90% of total polyphenols, catechol represents .apprx.4%, flavonols .apprx.0.8%, and phenolic acids .apprx.2% of the total polyphenols.

L6 ANSWER 24 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1970:131222 CAPLUS

DOCUMENT NUMBER: 72:131222

ORIGINAL REFERENCE NO.: 72:23497a,23500a

TITLE: Tannins in food technology with special reference to

brewing

AUTHOR(S): Reynolds, Thomas

CORPORATE SOURCE: Brew. Ind. Res. Found., Surrey, UK

SOURCE: Food Sci. Technol., Proc. Int. Congr., 1st (1969), Meeting Date 1962, 395-9. Editor(s):

Leitch, James Muil. Gordon and Breach Sci. Publ.: New

York, N. Y. CODEN: 15ZPAM

DOCUMENT TYPE: Conference LANGUAGE: English

AB Polyphenolic materials of a tanninlike nature are present in many plant materials. When these materials are processed in the production of food, the tannins often become altered and affect the quality of the finished product. Prominent among the polyphenols involved are the anthocyanogens (leucoanth ocyanins). These compound are involved, together with protein, in the formation of no nbiol. haze in beer, wine, and cider. The polyphenols undergo oxidation in the presence of air, probably catalyzed by heavy metal ions and the oxidized forms react with protein to form a pp t. This sequence occurs in many bottled beverages, such as beer, and is accelerated by unsuitable storage, the sale value of the beverage being thereby reduced. In beer, about 30-80%, depending upon the brewing process used, of the ant ocyanogens in the hopped wort, is derived from malt while the remainder or iginates from the hops. If beer is stored at a low temperature, a feature of the manufacture of lager, a

beer is stored at a low temperature, a feature of the manufacture of lage haze is

formed which redissolves on warming. If, however, this haze is filtered off from the chilled beer, subsequent formation of storage haze is retarded. Tannins are prominent constituents of other beverages and a series of polyphenols has been recognized in exts. of cocoa and tea. An example of deterioration of foods due to chemical change of the polyphenols is seen if soft fruits are preserved in a highly acidic environment when it frequently happens that the leucoanthocyanins are converted into anthocyanidin pigments which detract from the appearance of the product. This is particularly marked in the case of pears and gooseberries.

L6 ANSWER 25 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1970:120035 CAPLUS

DOCUMENT NUMBER: 72:120035

ORIGINAL REFERENCE NO.: 72:21579a,21582a

TITLE: Polymerization of polyhydric phenols

AUTHOR(S): Jerumanis, J.

CORPORATE SOURCE: Sect. Brass., Univ. Louvain, Louvain, Belg.

Bulletin de l'Association Royale des Anciens Etudiants SOURCE:

en Brasserie de l'Universite de Louvain (1969

), 65(4), 169-90

CODEN: BAEBA2; ISSN: 0365-8775

DOCUMENT TYPE: Journal; General Review

French LANGUAGE:

The polymerization index of anthocyanogens in hops, malt, barley

exts. and beer was determined and the polymerization mechanism of these

polyphenols was discussed. The amount of polyphenols in

beer was determined as a function of temperature and air content in the

review with 46 refs. on polyphenols polymerization is included.

ANSWER 26 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1970:53708 CAPLUS

DOCUMENT NUMBER: 72:53708 ORIGINAL REFERENCE NO.: 72:9815a,9818a

Determination of polyphenols in hops TITLE:

, hop extracts, barley, and malt

AUTHOR(S): Jerumanis, J.

CORPORATE SOURCE: Univ. Louvain, Louvain, Belg.

SOURCE: Bulletin de l'Association Royale des Anciens Etudiants

en Brasserie de l'Universite de Louvain (1969

), 65(3), 113-31

CODEN: BAEBA2; ISSN: 0365-8775

DOCUMENT TYPE: Journal LANGUAGE: French

Conventional hot water extraction of hops or extraction with EtOH, MeOH,

or Me2CO gave a low content of polyphenols in the ext.

as compared with extraction with 25% HCONMe2. Thus, certain samples of

hops contained 8-9% polyphenols (dry basis), while hot

water exts. of these contained 4-6%. The alcs. and Me2CO gave

higher results than hot water but not as high as HCONMe2. The use of HCONMe2 has the advantages of more complete extraction and preventing

oxidation of

the polyphenols during anal. Com. hop exts. were generally low in polyphenols, some samples containing none at all. The total polyphenol content of barleys and malts varied between 0.31 and 0.49% (dry basis). The polyphenol content of the malt is not necessarily related to that of the barley from which it was made. Variations in germination may account for this. The polyphenol content of several varieties of barley is given. Different samples of the same variety of barley may show varying polyphenol contents. The polyphenol content of barley and malt is only about 5% that of the hops. There appear to be some hop varieties that are generally high in polyphenols; other varieties are low. Storage of the hops from one year to another does not lower the polyphenol content.

ANSWER 27 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

1968:485422 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 69:85422

ORIGINAL REFERENCE NO.: 69:15947a,15950a

TITLE: Results of hop research and their application in

practice

AUTHOR(S): Schur, F.

CORPORATE SOURCE: Versuchssta. Schweiz Brau., Zurich, Switz. SOURCE: Schweizer Brauerei-Rundschau (1968), 79(7),

180 - 4

CODEN: SWBRA2; ISSN: 0036-7311

Journal DOCUMENT TYPE:

LANGUAGE: German

The chemical composition of  $\alpha$ -acids of hops, their stability during storage, and their influence on the bitter taste of beer were studied. Hops polyphenols, their influence on the stability of beer, essential oils of hops, their influence on the flavor of beer, and their enzymic modifications upon storage were also studied. The replacement of hops by hop exts. depends upon preisomerization, that is the isomerization of  $\beta$ -acids into bitter  $\alpha$ -acids. This could be achieved by uv irradiation. The chemical composition of hop exts. is influenced by the nature of the extraction solvent. MeOH extracted less essential oils (71% of total) than MeCl (94%). Gas chromatog. expts. showed a decrease in the most volatile and an increase in the less volatile components of the essential oils of preisomerized hop exts. Upon cold addition of hops, only 60% of the total isohumulone content of hops was found in beer. This was not due to the pH of beer but to the reaction of isohumulone with SH-containing substances of beer. 24 references.

L6 ANSWER 28 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1967:431539 CAPLUS

DOCUMENT NUMBER: 67:31539
ORIGINAL REFERENCE NO.: 67:5923a,5926a

TITLE: Process of wort boiling

AUTHOR(S): De Clerck, Jean CORPORATE SOURCE: Univ. Louvain, Belg.

SOURCE: Brewers Digest (1967), 42(3), 96-9

CODEN: BRDGAT; ISSN: 0006-971X

DOCUMENT TYPE: Journal LANGUAGE: English

Wort boiling stabilizes the wort by inactivating the enzymes and exts. valuable substances from the hops.  $\alpha ext{-Acids}$ and humulones are converted to isohumulones during wort boiling. Under the best conditions the yield of isohumulones is not greater than 65%. Hulupones (oxidation products of  $\beta$ -acids or lupulones), humililones (oxidation products of  $\alpha$ -acids), and a  $\delta$  component of hard resins yield a bitter taste less than that of isohumulones. They are completely soluble but are important only in old and oxidized hops. Two major factors for the coaqulation of proteins are pH and intensity of boiling. There is a decrease in wort pH during wort boiling as a result of liberation of H ions, precipitation of Ca3(PO4)2, and formation of melanoidins (which show acid reaction). Boiling leads to strong wort agitation thereby hastening protein coaqulation. With properly adjusted pH, an excellent break formation is achieved when evaporation rate is about 6%/hr. Essential oils undergo transformation during kettle boiling. They tend to form resins in the wort, giving rise to an acrid taste. Vacuum treatment of hops improves beer flavor by eliminating a major portion of the essential oils. There is marked increase in color during wort boiling due principally to the formation of melanoidins and oxidation of polyphenols. Oxidation improves beer stability. Addition of reducing agents causes an increase of 6-10% of iso-humulones. Stability of beer foam is improved by wort boiling. In the majority of cases hops are added to wort in several portions during the process of boiling.

L6 ANSWER 29 OF 29 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1955:58205 CAPLUS

DOCUMENT NUMBER: 49:58205 ORIGINAL REFERENCE NO.: 49:11236d-e

TITLE: Study of the tannins and polyphenols of

hops

AUTHOR(S): Vancraenenbroeck, R.; Lontie, R.

SOURCE: Bulletin de l'Association des Anciens Etudiants en

Brasserie de l'Universite de Louvain (1955),

51, 1-14

CODEN: BUAVAC; ISSN: 0366-3965

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

AB Expts. were conducted on the separation and analysis of hops based on chromatography and countercurrent distribution. The bitter resins were removed by continuous extraction with C6H6 followed by extraction of the polyphenols with 75% acetone. The acetone ext. was analyzed by partition chromatography. The presence of flavonols and leucoanthocyanins was confirmed.

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          2343 "HUMULUS"
          1432 "LUPULUS"
       1756191 "L"
          8032 "HOP"
          5381 "HOPS"
         10603 "HOP"
                ("HOP" OR "HOPS")
           509 "BRACT"
           943 "BRACTS"
          1242 "BRACT"
                 ("BRACT" OR "BRACTS")
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         17636 "BITTER"
                 ("BITTER" OR "BITTERS")
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            134 S L2 AND POLYPHENOLS
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             50 S L3 AND EXTRACT
L5
              0 S L4 AND PY, =2003
L6
             29 S L4 AND PY<=2003
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     FILE 'CAPLUS' ENTERED AT 15:27:42 ON 30 APR 2009
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=> s KANDA T?/AU
L9
          1828 KANDA T?/AU
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L10 ANSWER 1 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

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ACCESSION NUMBER: 2008:1041075 CAPLUS

DOCUMENT NUMBER: 150:90051

AUTHOR(S):

SOURCE:

TITLE: Inflammatory responses of gingival epithelial cells

stimulated with Porphyromonas gingivalis vesicles are

inhibited by hop-associated polyphenols Kou, Yurong; Inaba, Hiroaki; Kato, Takahiro;

Tagashira, Motoyuki; Honma, Daiki; Kanda,
Tomomasa; Ohtake, Yasuyuki; Amano, Atsuo

CORPORATE SOURCE: Department of Oral Frontier Biology, Osaka University

Graduate School of Dentistry, Suita-Osaka, Japan Journal of Periodontology (2008), 79(1), 174-180

Journal of Periodontology (2008), 79(1), 174-180

CODEN: JOPRAJ; ISSN: 0022-3492
HER: American Academy of Periodontology

PUBLISHER: American Academy of Periodontol DOCUMENT TYPE: Journal

DOCUMENT TYPE: Journal LANGUAGE: English

Periodontitis is induced by an imbalance between bacterial virulence and host defense ability. Porphyromonas gingivalis, a predominant periodontal pathogen, triggers a series of host inflammatory responses that aggravate the destruction of periodontium. Thus, anti-inflammatory reagents are considered desirable for effective periodontal therapy. In the present study, we examined the inhibitory effects of hop bract polyphenol (HBP) on cellular inflammatory responses induced by P. gingivalis membrane vesicles. Immortalized human gingival epithelial cells were stimulated with P. gingivalis membrane vesicles, and the effects of HBP on  $\ensuremath{\mathsf{mRNA}}$ expression of cyclooxygenase (COX)-2, interleukin (IL)-6 and -8, and matrix metalloproteinase (MMP)-1 and -3 were examined using real-time reverse transcription-polymerase chain reaction. HBP inhibited the mRNA expression of COX-2, IL-6 and -8, and MMP-1 and -3 in a dose-dependent manner, whereas epigallocatechin gallate (a control polyphenol) inhibited COX-2 mRNA expression only. Following further fractionation of HBP to identify the effective components,

 $2-[(2-\text{methylpropanoyl})-\text{phloroglucinol}]1-O-\beta-D-glucopyranoside (MPPG)$  was identified as a significant anti-inflammatory element that completely inhibited the inflammatory mRNA induction. Kaempferol  $3-O-\beta$ -glucopyranoside (astragalin) also was found to have anti-inflammatory effects. HBP is suggested to be a potent inhibitor of cellular inflammatory responses induced by P. gingivalis vesicles. Further, MPPG and astragalin, identified here as effective components of HBP, also may be useful for the prevention and/or attenuation of

periodontitis.

REFERENCE COUNT: 37 THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 2 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:912244 CAPLUS

DOCUMENT NUMBER: 149:168033

TITLE:  $2-acylphloroglucinol-4,6-di-C-\beta-D-glucopyranoside$ 

derivs. from Humulus lupulus as antioxidants

INVENTOR(S): Honma, Hiroki; Tagashira, Motoyuki;

Kanda, Tomomasa

PATENT ASSIGNEE(S): Asahi Breweries, Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 11pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

----JP 2008174458 A 20080731 JP 2007-7030 20070116

PRIORITY APPLN. INFO.: JP 2007-7030 20070116

OTHER SOURCE(S): MARPAT 149:168033

2-Acylphloroglucinol-4,6-di-C- $\beta$ -D-glucopyranoside derivs., e.g. 2-(3-methylbutyryl) phloroglucinol-4,6-di-C- $\beta$ -D-glucopyranoside, are claimed as antioxidants, health foods, and cosmetics. Formulation examples of capsules, granules, injections, lotions, and health foods were given.

L10 ANSWER 3 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:912243 CAPLUS

DOCUMENT NUMBER: 149:207994

Antioxidants comprising phloroglucinol glycoside TITLE:

INVENTOR(S): Honma, Hiroki; Tagashira, Motoyuki;

Kanda, Tomomasa

Asahi Breweries, Ltd., Japan PATENT ASSIGNEE(S): Jpn. Kokai Tokkyo Koho, 10pp. SOURCE:

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_\_ \_\_\_\_ JP 2008174457 A 20080731 JP 2007-7029 JP 2007-7029 20070116 PRIORITY APPLN. INFO.: This invention relates to antioxidants, therapeutic agents, cosmetics, and

foods/beverages comprising 2-(2-methylpropanoy1)phloroglucinol-1,5-di-O- $\beta$ -D-glucopyranoside (I) extracted from hop.

L10 ANSWER 4 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:632770 CAPLUS

DOCUMENT NUMBER: 149:866

Identification of hop polyphenolic components which TITLE:

inhibit prostaglandin E2 production by gingival

epithelial cells stimulated with periodontal pathogen

Inaba, Hiroaki; Tagashira, Motoyuki; Honma, AUTHOR(S):

Daiki; Kanda, Tomomasa; Kou, Yurong; Ohtake,

Yasuyuki; Amano, Atsuo

CORPORATE SOURCE: Department of Oral Frontier Biology, Osaka University

Graduate School of Dentistry, 1-8 Yamadaoka, Suita,

Osaka, 565-0871, Japan

SOURCE: Biological & Pharmaceutical Bulletin (2008), 31(3),

527-530

CODEN: BPBLEO; ISSN: 0918-6158 Pharmaceutical Society of Japan

DOCUMENT TYPE: Journal

LANGUAGE: English

PUBLISHER:

Chronic marginal periodontitis is a destructive inflammatory disease AB caused by an imbalance between bacterial virulence and host defense ability, resulting in eventual tooth exfoliation. Porphyromonas gingivalis, a major periodontal pathogen, triggers a series of cellular inflammatory responses including the production of prostaglandin E2 (PGE2), which causes periodontal destruction; thus, anti-inflammatory reagents are considered beneficial for periodontal therapy. In the present study, we examined whether hop- and apple-derived polyphenols (HBP and ACT, resp.) inhibit PGE2 production by human gingival epithelial (HGE) cells stimulated with P. gingivalis components. HGE cells were stimulated with P. gingivalis membrane vesicles, and the effects of HBP, ACT and epigallocatechin gallate (EGCg) on PGE2 production by HGE cells were evaluated using an ELISA. HBP and EGCg significantly inhibited PGE2 production, whereas

ACT did not. By further fractionation steps of HBP to identify the effective components, 3 components of HBP,  $2-[(2-methylpropanoyl)-phloroglucinol]1-O-\beta-D-glucopyranoside (MPPG),$ quercetin 3-0- $\beta$ -D-glucopyranoside (isoquercitrin), and kaempferol  $3-0-\beta$ -glucopyranoside (astragalin), were found to be elements which significantly inhibited cellular PGE2 production These results suggest that HBP is a potent inhibitor of cellular PGE2 production induced by P. gingivalis, and HBP may be useful for the prevention and attenuation of

REFERENCE COUNT: 27 THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 5 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:286262 CAPLUS

DOCUMENT NUMBER: 148:315151

TITLE: Method for production of hop preparation, hop

preparation, antiinflammatory agent, food/beverage,

and oral product

INVENTOR(S): Inaba, Hiroaki; Honma, Daiki; Tagashira,

Motoyuki; Kanda, Tomomasa; Amano, Atsuo

Asahi Breweries, Ltd., Japan PATENT ASSIGNEE(S):

SOURCE: PCT Int. Appl., 27pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

periodontitis.

PATENT INFORMATION:

PA:	PATENT NO.					D	DATE			APPL	ICAT		DATE					
WO 2008026473					A1	_	20080306		•	WO 2007-JP66148						20070821		
	W:	ΑE,	AG,	AL,	AM,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BH,	BR,	BW,	BY,	BZ,	CA,	
		CH,	CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DO,	DZ,	EC,	EE,	EG,	ES,	FΙ,	
		GB,	GD,	GE,	GH,	GM,	GT,	HN,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	
		KM,	KN,	KP,	KR,	KΖ,	LA,	LC,	LK,	LR,	LS,	LT,	LU,	LY,	MA,	MD,	ME,	
		MG,	MK,	MN,	MW,	MX,	MY,	MZ,	NA,	NG,	NΙ,	NO,	NΖ,	OM,	PG,	PH,	PL,	
		PT,	RO,	RS,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SM,	SV,	SY,	ΤJ,	TM,	TN,	
		TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	ZA,	ZM,	ZW					
	RW:	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FΙ,	FR,	GB,	GR,	HU,	IE,	
		IS,	ΙΤ,	LT,	LU,	LV,	MC,	MT,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,	BF,	
		ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	G₩,	ML,	MR,	ΝE,	SN,	TD,	ΤG,	BW,	
		GH,	GM,	KΕ,	LS,	MW,	MΖ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	ΑM,	ΑZ,	
		BY,	KG,	KΖ,	MD,	RU,	ТJ,	TM										

PRIORITY APPLN. INFO.:

JP 2006-236774 A 20060831

Provided is a method for producing a preparation which can be used for the prevention/treatment of an inflammatory disease including gingivitis caused by Porphyromonas gingivalis. The method comprises the following steps (1) to (3): (1) adjusting the pH value of a liquid solution containing a polyphenol fraction produced from a hop bract to 6 to 7, and passing the liquid solution through a styrene-divinylbenzene resin to cause the adsorption of components including a useful substance onto the resin; (2) washing the resin obtained in the step (1) with a 30-60% aqueous ethanol solution to elute out and remove an undesired substance from the components adsorbed on the resin; and (3) eluting out components including the useful substance from the resin produced in the step (2) with a 70% or higher aqueous ethanol solution

or ethanol and producing a preparation by using the resulting elution fraction. THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 17 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ACCESSION NUMBER: 2007:771458 CAPLUS

DOCUMENT NUMBER: 147:405163

TITLE: Safety evaluation of polyphenols extracted from hop

bracts

AUTHOR(S): Nagasako-Akazome, Yoko; Honma, Daiki; Tagashira,

Motoyuki; Kanda, Tomomasa; Yasue,

Masaaki; Ohtake, Yasuyuki

CORPORATE SOURCE: Fundamental Research Laboratory, Asahi Breweries,

Ltd., Moriya-shi, Ibaraki, 302-0106, Japan

SOURCE: Food and Chemical Toxicology (2007), 45(8), 1383-1392

CODEN: FCTOD7; ISSN: 0278-6915

PUBLISHER: Elsevier Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Hop bract polyphenols contain polyphenols as promising functional ingredients. To assess the safety of topical hop bract polyphenols, Hopsphenon we examined acute, 14-day, 28-day and 90-day toxicity tests in rats, and mutagenicity tests using Ames test and micronucleus test in mice. The acute, 14-day, 28-day and 90-day toxicity tests revealed that Hopsphenon produced no symptoms of significant injury. The LD of hop bract polyphenols is greater than 2000 mg/kg. The Ames test in the absence of S9 mix for TA98 and in the presence of S9 mix for TA1537 revealed that Hopsphenon had slight mutagenicity at a high dose of 5000  $\mu \text{g}/\text{plate}$ ; however, in the micronucleus test, Hopsphenon was neg. These tests demonstrated that hop bract polyphenols are safe and do not cause any detrimental effects in vivo under the conditions investigated in this study.

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 7 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2007:200713 CAPLUS

DOCUMENT NUMBER: 146:212295

TITLE: Oral compositions containing cariostatic agents and

gums

INVENTOR(S): Akazome, Yoko; Tagashira, Motoyuki;

Kanda, Tomomasa; Hirai, Nobuaki Asahi Breweries, Ltd., Japan

SOURCE: PCT Int. Appl., 26pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT ASSIGNEE(S):

PA.	PATENT NO.					KIND DATE			APPLICATION NO.							DATE		
WO	WO 2007020830				A1 200702			0222	,	WO 2		20060808						
	W:	ΑE,	AG,	AL,	AM,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,	
		CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FΙ,	GB,	GD,	
		GE,	GH,	GM,	HN,	HR,	HU,	ID,	IL,	IN,	IS,	ΚE,	KG,	KM,	KN,	KP,	KR,	
		KΖ,	LA,	LC,	LK,	LR,	LS,	LT,	LU,	LV,	LY,	MA,	MD,	MG,	MK,	MN,	MW,	
		MX,	MZ,	NA,	NG,	NΙ,	NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RS,	RU,	SC,	
		SD,	SE,	SG,	SK,	SL,	SM,	SY,	ТJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	
		UZ,	VC,	VN,	ZA,	ZM,	ZW											
	RW:	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FΙ,	FR,	GB,	GR,	HU,	ΙE,	
		IS,	ΙT,	LT,	LU,	LV,	MC,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,	BF,	ВJ,	
		CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	ΤG,	BW,	GH,	
		GM,	KΕ,	LS,	MW,	MZ,	NΑ,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	ΑZ,	BY,	
		KG,	KΖ,	MD,	RU,	ТJ,	TM											
JP 2007051096					Α	20070301			1	JP 2	005-	2376	48		20050818			

PRIORITY APPLN. INFO.: JP 2005-237648 A 20050818

It is intended to provide an oral composition, which is safe and effective even if it is continuously used and is characterized by showing a cariostatic action, an anti-periodontal disease action and an anti-halitosis action. The oral composition showing a cariostatic action, an anti-periodontal disease action and an anti-halitosis action is characterized by containing components of (a) proanthocyanidins as a cariostatic material and/or an anti-periodontal disease material and (b) a soluble viscosity-enhancing polysaccharide as a gumming agent as essential components.

REFERENCE COUNT: 20 THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 8 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

2007:164217 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 146:417566

TITLE: Retention behavior of oligomeric proanthocyanidins in

hydrophilic interaction chromatography

Yanagida, Akio; Murao, Hirokazu; Ohnishi-Kameyama, AUTHOR(S):

Mayumi; Yamakawa, Yutaka; Shoji, Atsushi; Tagashira, Motoyuki; Kanda, Tomomasa ; Shindo, Heisaburo; Shibusawa, Yoichi

CORPORATE SOURCE: Division of Structural Biology and Analytical Science,

> School of Pharmacy, Tokyo University of Pharmacy and Life Science, 1432-1 Horinouchi, Hachioji, Tokyo,

192-0392, Japan

SOURCE: Journal of Chromatography, A (2007), 1143(1-2),

153-161

CODEN: JCRAEY; ISSN: 0021-9673

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

A novel method was developed for the separation of proanthocyanidins (PAs; oligomeric flavan-3-ols) by hydrophilic interaction chromatog. (HILIC) using an amide-silica column eluting with an aqueous acetonitrile mobile phase. The best separation was achieved with a linear gradient elution of acetonitrile-water at ratios of 9:1 to 5:5 (volume/volume) for 60 min at a flow rate of  $1.0~\mathrm{mL/min}$ . Under these HPLC conditions, a mixture of natural oligomeric PAs (from apple) was separated according to d.p. (DP) up to decamers. The DP of each separated oligomer was confirmed by LC/electrospray ionization MS. In further HILIC separation studies of 15 different flavan-3-ol and oligomeric PA (up to pentamer) stds. with an isocratic elution of acetonitrile-water (84:16), a high correlation was observed between the logarithm of retention factors (log k) and the number of hydroxyl groups in their structures. The coefficient of this correlation (r2 = 0.9501) was larger than the coefficient (r2 = 0.7949) obtained from the correlation between log k and log Po/w values. These data reveal that two effects, i.e. hydrogen bonding between the carbamoyl terminal on the column and the hydroxyl group of solute oligomer and hydrophilicity based on the high-order structure of oligomeric PAs, corporately contribute to the separation, but the hydrogen bonding effect is predominant in the authors' HILIC separation mode.

THERE ARE 36 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 36 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 9 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2006:78441 CAPLUS

DOCUMENT NUMBER: 145:937

TITLE: Apple- and hop-polyphenols protect periodontal

> ligament cells stimulated with enamel matrix derivative from Porphyromonas gingivalis

Inaba, Hiroaki; Tagashira, Motoyuki; AUTHOR(S):

Kanda, Tomomasa; Ohno, Takashi; Kawai, Shinji;

Amano, Atsuo

CORPORATE SOURCE: Department of Oral Frontier Biology, Osaka University

Graduate School of Dentistry, Osaka, Japan

SOURCE: Journal of Periodontology (2005), 76(12), 2223-2229

CODEN: JOPRAJ; ISSN: 0022-3492

PUBLISHER: American Academy of Periodontology

DOCUMENT TYPE: Journal LANGUAGE: English

Enamel matrix derivative (EMD) is a tissue regenerative agent used clin. as an AΒ adjunct to periodontal surgery. It was previously demonstrated that Porphyromonas gingivalis, a periodontal pathogen, significantly diminished the efficacy of EMD with periodontal ligament (PDL) cells through the proteolytic actions of Arg- and Lys-gingipains (Rgp and Kgp). Thus, antiproteolytic supplements are considered clin. desirable for effective periodontal regenerative therapies. In the present study, we examined apple- (AP) and hop-polyphenols to determine their ability to protect EMD-stimulated PDL cells from P. gingivalis. AP, apple condensed tannin (ACT), hop bract polyphenol (HBP), high and low mol. weight fractions of HBP (HMW-HBP and LMW-HBP), and epigallocatechin gallate (EGCg) were used. PDL cells were grown on EMD-coated dishes and infected with P. gingivalis, and cellular migration and proliferation were evaluated with an in vitro assay of wound healing assay in the presence or absence of the polyphenols. Each polyphenol significantly enhanced the viability of PDL cells infected with P. gingivalis, whereas only EGCg demonstrated cytotoxicity. Further, all polyphenols significantly inhibited Rgp activity, with AP, ACT, and HBP more effective toward Kgp. P. gingivalis markedly diminished the migration and proliferation of EMD-stimulated PDL cells, whereas the addition of AP, ACT, HBP, and HMW-HBP significantly protected the cells from bacterial cytotoxicity. In contrast, EGCg and LMW-HBP did not show protective effects. These results suggest that AP, ACT, AP, HBP, and HMW-HBP protect EMD-stimulated PDL cells from P. gingivalis and may be therapeutically useful supplements for EMD therapy.

REFERENCE COUNT: 48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 10 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2005:1075625 CAPLUS

DOCUMENT NUMBER: 143:352894

TITLE: Periodontal ligament-protecting agents containing

proanthocyanidin-like polyphenols Inaba, Hiroaki; Tagashira, Motoyuki;

Kanda, Tomomasa

PATENT ASSIGNEE(S): Asahi Breweries, Ltd., Japan; Inaba, Hiroaki;

Tagashira, Motoyuki; Kanda, Tomomasa; Amano, Atsuo

SOURCE: PCT Int. Appl., 21 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

INVENTOR(S):

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2005092327	A1	20051006	WO 2005-JP3166	20050225
W: AE, AG,	L, AM, AT,	, AU, AZ, BA,	BB, BG, BR, BW,	BY, BZ, CA, CH,
CN, CO,	R, CU, CZ,	, DE, DK, DM,	DZ, EC, EE, EG,	ES, FI, GB, GD,
GE, GH,	M, HR, HU,	, ID, IL, IN,	IS, JP, KE, KG,	KP, KR, KZ, LC,
LK, LR,	S, LT, LU,	, LV, MA, MD,	MG, MK, MN, MW,	MX, MZ, NA, NI,
NO, NZ,	M, PG, PH,	, PL, PT, RO,	RU, SC, SD, SE,	SG, SK, SL, SM,
SY, TJ,	M, TN, TR,	, TT, TZ, UA,	UG, US, UZ, VC,	VN, YU, ZA, ZM, ZW
RW: BW, GH,	M, KE, LS,	, MW, MZ, NA,	SD, SL, SZ, TZ,	UG, ZM, ZW, AM,

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AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT,
             RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML,
             MR, NE, SN, TD, TG
     CA 2520166
                                20051006
                                          CA 2005-2520166
                          Α1
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     AU 2005203558
                          Α1
                                20051013
                                            AU 2005-203558
                                                                    20050225
     CN 1764448
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                                20060426
                                            CN 2005-80000052
                                                                    20050225
                                20061206
                                            EP 2005-710716
     EP 1728509
                          Α1
                                                                    20050225
            AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
             IS, IT, LI, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR
                                                                    20050810
     KR 2006034624
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                                            KR 2005-714687
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     KR 821694
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                                20080411
     US 20060165609
                          Α1
                                20060727
                                            US 2005-545412
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     US 20070092456
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                                20070426
                                            US 2006-640301
                                                                    20061218
                                            KR 2007-718773
     KR 2007097581
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                          Α
PRIORITY APPLN. INFO.:
                                             JP 2004-91098
                                                                 A 20040326
                                            WO 2005-JP3166
                                                                 W 20050225
                                                                 A3 20050810
                                             KR 2005-714687
                                             US 2005-545412
                                                                 A1 20050812
AΒ
     It is intended to provide a periodontal ligament-protecting agent which
     prevents failures in the periodontal tissue (in particular, periodontal
     ligament) caused by Porphyromonas gingivalis, and an oral preparation, a food
     or a drink having effects of preventing and treating diseases relating to
     damage in periodontal ligament which contains this periodontal
     ligament-protecting agent. Namely, a periodontal ligament-protecting
     agent which is a proanthocyanidine having an effect of relieving toxicity
     of Porphyromonas gingivalis to periodontal ligament (preferably a
     proanthocyanidine originating in immature apple fruit or hop bract); and
     an oral preparation, a food or a drink containing this periodontal
     ligament-protecting agent as the active ingredient.
REFERENCE COUNT:
                         78
                               THERE ARE 78 CITED REFERENCES AVAILABLE FOR THIS
                               RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
L10 ANSWER 11 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                         2005:965910 CAPLUS
DOCUMENT NUMBER:
                         143:228201
TITLE:
                         Analysis of polyphenols from hop bract region using
AUTHOR(S):
                         Kurumatani, Masami; Fujita, Rumi; Tagashira,
                         Motoyuki; Shoji, Toshihiko; Kanda,
                         Tomomasa; Ikeda, Mitsuo; Shoji, Atsushi;
                         Yanagida, Akio; Shibusawa, Yoichi; Shindo, Heisaburo;
                         Ito, Yoichiro
CORPORATE SOURCE:
                         Fundamental Research Laboratory, Asahi Breweries,
                         Ltd., Moriya, Ibaraki, Japan
                         Journal of Liquid Chromatography & Related
SOURCE:
                         Technologies (2005), 28(12-13), 1971-1983
                         CODEN: JLCTFC; ISSN: 1082-6076
                         Taylor & Francis, Inc.
PUBLISHER:
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
     Polyphenols derived from hop (Humulus lupulus L.) bract region (HBP) can
AΒ
     be used as food materials, thereby preventing dental caries. Chemical
     details of the active substances need to be elucidated. The polyphenols
     from hop bract (HBP) region were purified by countercurrent chromatog.
     (CCC). The fractions were analyzed by high-performance size-exclusion
     chromatog. (HPSEC) and reversed phase high-performance liquid chromatog.
     (RP-HPLC). From HBP fractions by HPSEC, some low-mol.- weight polyphenols
     (glycosides of flavonoids, catechins, and proanthocyanidins) were
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identified by RP-HPLC. However, a very hydrophilic fraction was found to have the most potent cavity-preventive activity, but it showed no peak in

its RP-HPLC chromatogram (absence of small polyphenols). HPSEC anal. showed that the major components of this fraction were high-mol. weight substances, which were supposed to be proanthocyanidins, consisting of

approx. 22 catechin units in its structure.

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 12 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2004:965021 CAPLUS

DOCUMENT NUMBER: 141:384030

TITLE: Material for inhibiting enamel decalcification

INVENTOR(S): Imai, Susumu; Tagashira, Motoyuki;

Kanda, Tomomasa; Nishizawa, Toshiki; Hanada,

Nobuhiro

PATENT ASSIGNEE(S): Asahi Breweries Ltd., Japan

SOURCE: PCT Int. Appl., 23 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA:	TENT :	NO.			KIN	D	DATE			APPL	ICAT	ION I	NO.		DATE				
WO	2004	0961	 65		A1	_	 2004	1111	,	WO 2004-JP6465						20040430			
	W:	ΑE,	AG,	AL,	AM,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,		
		CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,		
		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	KP,	KR,	KΖ,	LC,		
		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MΖ,	NΑ,	NΙ,		
		NO,	NΖ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,		
		ΤJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UΖ,	VC,	VN,	YU,	ZA,	ZM,	ZW		
	RW:	BW,	GH,	GM,	KΕ,	LS,	MW,	${ m MZ}$ ,	NA,	SD,	SL,	SZ,	${\sf TZ}$ ,	UG,	ZM,	ZW,	AM,		
								ТJ,											
								HU,											
		•			BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,		
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	2524				A1			1111								0040			
EP	1621				A1			0201											
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	8109	-						0310			005-					0051	-		
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US 20080003186 DRITY APPLN. INFO.:					AI		2008	0103			007- 003-					0070 0030			
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It is intended to provide an effective cariostatic material which inhibits dental plaque formation as well as onset of dental caries. A material for inhibiting enamel decalcification containing, as the active ingredient, a proanthocyanidin-like polyphenol originating in hop bract or immature apple, which effectively inhibits not only dental plaque formation but also the dental caries process including proliferation of bacteria, formation of acids by the bacteria and enamel decalcification. Also, foods, drinks and oral care goods with the use of the above substance as an enamel decalcification inhibitor are provided. An enamel decalcification inhibitor was prepared from immature apple fruit extract The obtained enamel decalcification inhibitor was combined at 0.005 % with other ingredients to 100 % to give a tooth paste.

REFERENCE COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS L10 ANSWER 13 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2004:758823 CAPLUS

DOCUMENT NUMBER: 141:276589

Method for separating polyphenols TITLE:

INVENTOR(S): Yanagida, Akio; Shibusawa, Yoichi; Kamifuji,

Heisaburo; Tagashira, Motoyuki; Kanda,

Tomomasa

PATENT ASSIGNEE(S): Asahi Breweries, Ltd, Japan SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004256481 PRIORITY APPLN. INFO.:	A		JP 2003-51143 JP 2003-51143	20030227 20030227

AΒ The title method comprises using high-speed counter-current chromatog. (liquid-liquid partition chromatog.) and a two-phase solvent system (consisting of an ether, acetonitrile, or an alc., water or acidic aqueous solution) in which the upper layer (or lower layer) is the stationary phase and the lower layer (or upper layer) is the mobile phase. The title method is used for quality control for food, pharmaceuticals, and cosmetics containing polyphenols.

L10 ANSWER 14 OF 14 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2004:515517 CAPLUS

141:33848 DOCUMENT NUMBER:

TITLE: Process for producing hop glume polyphenols

INVENTOR(S): Tagashira, Motoyuki; Kanda, Tomomasa

PATENT ASSIGNEE(S): Asahi Breweries, Ltd., Japan

SOURCE: PCT Int. Appl., 16 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA.	PATENT NO.					KIND DATE			APPLICATION NO.							DATE		
WO	2004	0528	98		A1		2004	0624	WO 2003-JP15959						20031212			
	W:	ΑU,	CN,	JP,	US													
	RW:	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK, E	Ε,	ES,	FI,	FR,	GB,	GR	, IE,	ΙΤ,	
		LU,	MC,	NL,	PT,	SE,	SK,	TR										
AU	20032	2890	63		A1		2004	0630	AU	20	03-	2890	63			20031	212	
AU	20032	2890	63		В2		2007	1018										
EP	15773	315			A1		2005	0921	EP	20	003-	7788	86			20031	212	
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB, G	R,	ΙT,	LI,	LU,	NL,	SE	, MC,	PT,	
		ΙE,	LT,	LV,	FI,	MK,	CY,	AL,	TR, B	G,	CZ,	EE,	SK					
CN	17262	221			A		2006	0125	CN	20	003-	8010	5813			20031	212	
US	20060	0251	760		A1		2006	1109	US	20	05-	5387	90			20050	610	
PRIORIT	Y APP	LN.	INFO	. :					JP	20	02-	3604	24		A	20021	212	
									WO	20	03-	JP15	959	,	W	20031	212	
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AB This invention provides a process for efficiently producing highly purified hop glume polyphenols using hop glume as the starting material; food, drinks, cosmetics and drugs containing hop glume polyphenol are disclosed. Namely, a process for producing hop polyphenols comprises

extracting hop glume with an aqueous alc. solution, concentrating the extract to give a residual

alc. concentration of 0.5 to 2% and then centrifuging and/or filtering the concentrate  $\,$ 

Formulations containing hop glume polyphenols are given.

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> FIL STNGUIDE COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 68.64 169.41 SINCE FILE DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) TOTAL ENTRY SESSION -11.48 CA SUBSCRIBER PRICE -35.26

FILE 'STNGUIDE' ENTERED AT 15:30:14 ON 30 APR 2009 USE IS SUBJECT TO THE TERMS OF YOUR CUSTOMER AGREEMENT COPYRIGHT (C) 2009 AMERICAN CHEMICAL SOCIETY (ACS)

FILE CONTAINS CURRENT INFORMATION.
LAST RELOADED: Apr 24, 2009 (20090424/UP).

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(FILE 'HOME' ENTERED AT 15:21:58 ON 30 APR 2009)

FILE 'CAPLUS' ENTERED AT 15:22:17 ON 30 APR 2009

L1 0 S HOPBRACT

L2 5382 S HOPS

L3 134 S L2 AND POLYPHENOLS

L4 50 S L3 AND EXTRACT

L5 0 S L4 AND PY,=2003

L6 29 S L4 AND PY<=2003

FILE 'STNGUIDE' ENTERED AT 15:24:27 ON 30 APR 2009

FILE 'CAPLUS' ENTERED AT 15:27:42 ON 30 APR 2009

L7 0 S "HUMULUS LUPULUS L" "HOP BRACT POLYPHENOLS" "HOP BITTER ACIDS

L8 122 S TAGASHIRA M?/AU

L9 1828 S KANDA T?/AU

L10 14 S L8 AND L9

FILE 'STNGUIDE' ENTERED AT 15:30:14 ON 30 APR 2009